

The Effect of Pulse Laser Ablation Method on the Morphology of Ta₂O₅ Nanoparticles

Irma Safira, Eko Hidayanto, and Ali Khumaeni*

Department of Physics, Faculty of Science and Mathematics, Diponegoro University, Jl. Prof. Suharto, SH, Tembalang, Semarang 50275, Indonesia

Corresponding email: khumaeni@fisika.fsm.undip.ac.id



Abstract — In this study, tantalum pentoxide (Ta₂O₅) nanoparticles were synthesized using the pulsed laser ablation method with deionized water (DIW) as liquid medium. The pulse laser ablation method produces nanoparticles with high purity due to its lack of reactant contamination. Ta₂O₅ nanoparticles were successfully synthesized with a UV-Vis wavelength of 293 nm. FESEM images produced excellent images with an average nanoparticle size of 16.92 nm.

Keywords— Tantalum Pentoxide Nanoparticles; Pulsed Laser Ablation; Deionized Water.

I. INTRODUCTION

Tantalum pentoxide (Ta₂O₅) nanoparticles have attracted significant attention in the fields of materials and nanotechnology research due to their unique characteristics. Ta₂O₅ nanomaterials are widely studied due to their excellent dielectric and electrochemical properties [1]. Ta₂O₅ nanoparticles have good biocompatibility, low ion release, corrosion resistance, low toxicity, and can be used for biomedical applications [2-4]. In addition, Ta O₅ nanoparticles have been widely applied in various fields, including orthopedics, electronics industry, energy, and catalysts [5-6].

Several researchers have conducted research on Ta₂O₅ nanoparticles with various different synthesis methods. Ta₂O₅ nanoparticles can be synthesized using the hydrogen arc plasma method conducted by Wang et al. (2004). The results of the study showed that tantalum nanoparticles have high activity and are easily oxidized, as confirmed by UV-Vis, XRD, TEM, and HRTEM analysis [7]. Streetwong et al. (2013) synthesized Ta₂O₅ nanoparticles using the sol-gel method. The synthesis results showed that Ta₂O₅ nanoparticles were most thermally stable at an optimum temperature of 700°C. In addition, Ta₂O₅ nanoparticles can show very good resistance in repeated use [8]. Nagaraju et al. (2019) synthesized Ta₂O₅ nanoparticles using the ultrasound-assisted synthesis method. The results showed that Ta₂O₅ nanoparticles have beneficial activity against antibacterial and anti-breast cancer [9].

In this study, we will be producing Ta₂O₅ nanoparticles using the pulse laser ablation method with deionized water as the liquid media. Ta₂O₅ nanoparticles were subsequently examined using UV-Vis and FESEM methods. This study aims to evaluate the effect of the pulsed laser ablation technique on the obtained Ta₂O₅ nanoparticles.

II. EXPERIMENTAL PROCEDURE

2.1 Materials

Tantalum plate (Ta) 99%, deionized water (DIW), Nd:YAG Laser (Q-smart 850 by Quantel).

2.2 Methods of Synthesis of Tantalum Nanoparticles

Figure 1 shows the synthesis method of Ta₂O₅ nanoparticles. Preparation was carried out before data collection. After cutting the Ta plate to 1.5 cm x 1.5 cm, we placed it on the bottom of a petri dish and added 10 ml of liquid medium, specifically deionized water. We used a pulse Nd:YAG laser for the synthesis of colloidal nanoparticles. During the synthesis process, a silver mirror deflected the laser beam towards a convex lens with a focal length of 10 cm, directing it towards the target at the bottom of the petri dish.

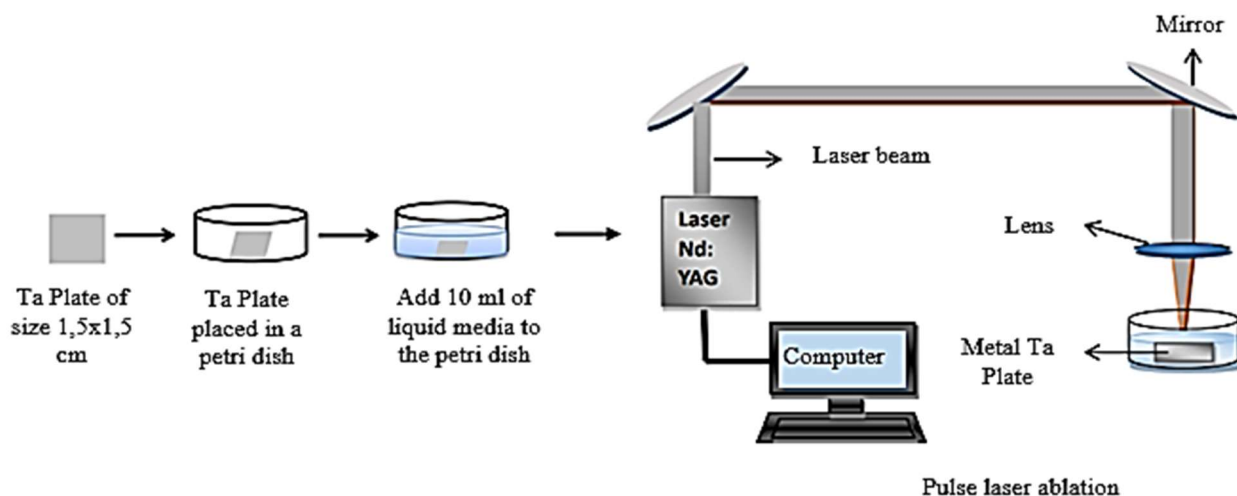


Figure 1 Synthesis method of Ta₂O₅ nanoparticles

2.3 Nanoparticles Characterization

Ultraviolet visible (UV-Vis; Shimadzu UVmini-1240) spectroscopy is used to determine the absorption of a material or sample. The sample is inserted into a quartz cuvette with as much as 2 ml. We use deionized water as the blank solution, utilizing the same cuvette for the analysis. Set the tool with an absorbance of around 3 AU and a wavelength between 200 and 700 nm. We analyze the morphology of nanoparticles using the Field Emission Scanning Electron Microscope (FESEM; JSM-IT 700 HR) and process the FESEM data using ImageJ software.

III. RESULTS AND DISCUSSION

Colloidal nanoparticles of Ta₂O₅ in deionized water are shown in Figure 2. The colloidal image of Ta₂O₅ nanoparticles before synthesis has a clear color, while after synthesis it has a gray color.

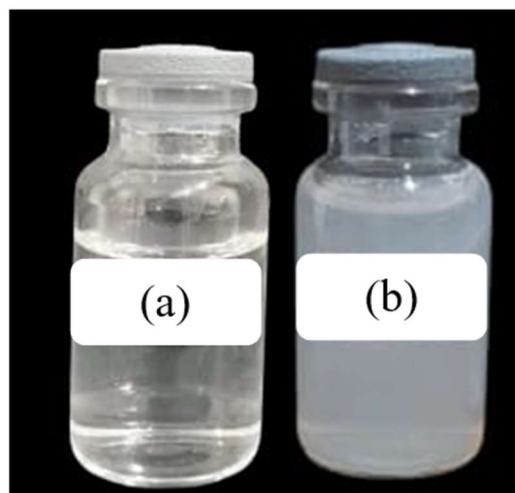


Figure 2 Ta₂O₅ nanoparticles in deionized water liquid media (a) before synthesis and (b) after synthesis

Figure 2 represents a color transition from clear before synthesis to gray post-synthesis, indicating the formation of Ta₂O₅ nanoparticles. According to a study by Gomes et al. (2018), the produced Ta₂O₅ nanoparticles have a gray hue [10]. The Ta₂O₅ nanoparticle colloid will then be studied by UV-Vis spectroscopy and field emission scanning electron microscopy (FESEM). The resulting colloid of Ta₂O₅ nanoparticles was studied using UV-Vis spectroscopy.

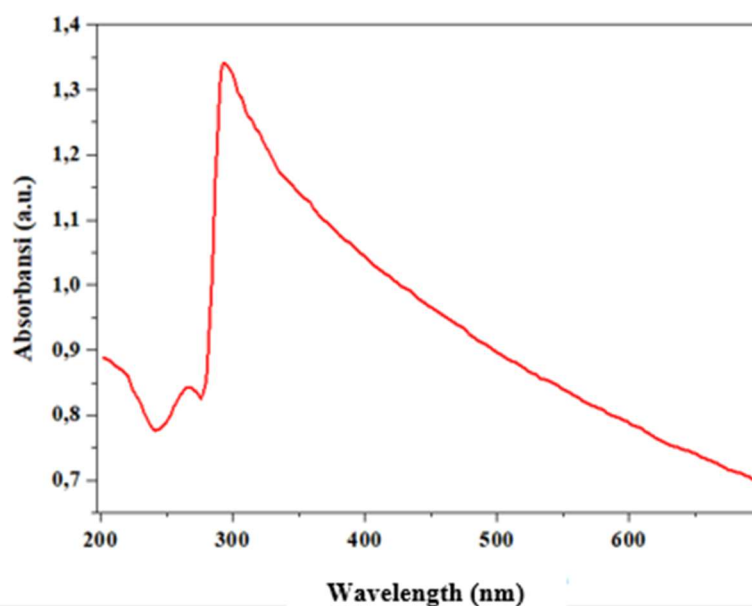


Figure 3 UV-Vis spectrum of Ta₂O₅ nanoparticles

Figure 3 shows the UV-Vis spectrum graph of the nanoparticle colloid that was made using the pulse laser ablation method. After synthesizing the nanoparticle colloid for 30 minutes, we observed an absorption peak at a wavelength of 293 nm, resulting in an absorption of 1.341. This study demonstrates the successful formation of Ta₂O₅ nanoparticles in the sample. Gomes et al. (2018) found that the solvothermal synthesis method, which involves the reaction of Ta(V) chloride with benzyl alcohol, produces an

absorption peak at a wavelength between 250 and 318 nm. Increasing the temperature increases the absorption value, resulting in a greater concentration of particles formed at the absorption height [10].

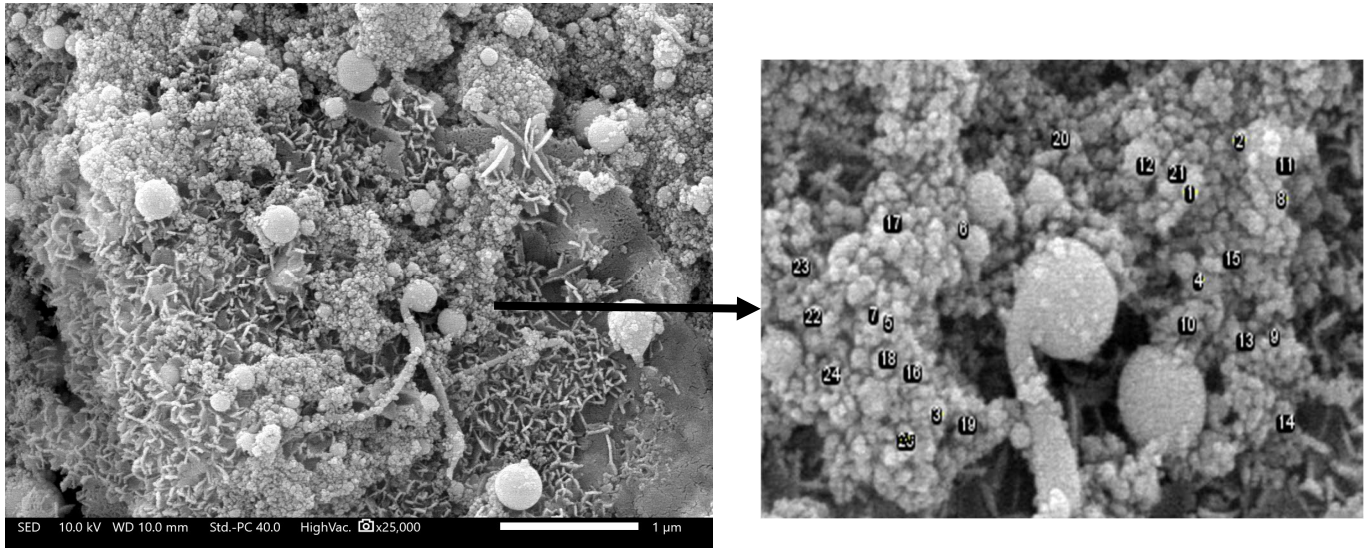
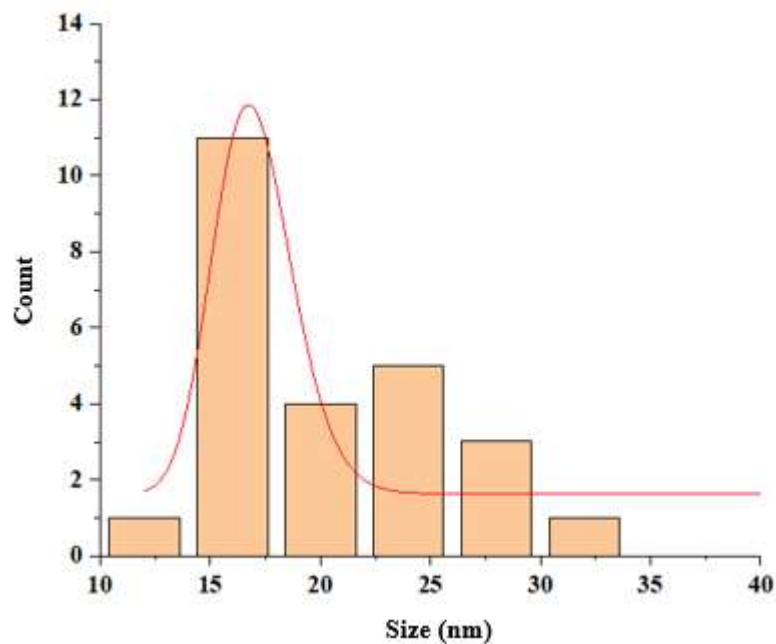


Figure 4 FESEM image Ta_2O_5 nanoparticles



Figs. 5. Size distribution of Ta_2O_5 nanoparticles

Furthermore, characterization analysis was carried out using *field electron scanning emission microscope* (FESEM). The FESEM image results in Figure 4 clearly illustrate the morphology of the sample, with a scale of 1 μm . When heated by the laser,

the Brownian motion effect causes the Ta₂O₅ nanoparticles that were made with deionized water to stick together. In addition, agglomeration can be caused by the liquid media used; although deionized water only contains a few ions in it, nanoparticle interactions can still occur through van der Waals forces. This leads to the attraction and self-agglomeration of the nanoparticles [11]. For the distribution of Ta₂O₅ nanoparticles, it was processed using ImageJ software, so that a histogram was obtained that describes the size distribution. We achieve this by measuring the diameter of the nanoparticles visible in the FESEM image. Figure 5 produces a histogram analysis of the size distribution of tantalum nanoparticles. The diameter of the nanoparticles ranges from 12.01 to 32.03 nm, with an average diameter of 16.92 nm.

In this study, the diameter of the nanoparticles was much smaller compared to previous studies on Ta₂O₅ nanoparticles. For example, Lu et al. (2021) found that using cathode glow discharge electrolysis (CGDE) at 500 volts to make Ta₂O₅ nanoparticles with a size of 325 nm was one way to do it [12]. This study shows that the use of the pulse laser ablation method produces a smaller size compared to other methods in the synthesis of Ta₂O₅ nanoparticles.

IV. CONCLUSIONS

Synthesis of Ta₂O₅ nanoparticles using the pulse laser ablation method has been successfully carried out. The results showed that the formed Ta₂O₅ nanoparticles had a nanoparticle diameter ranging from 12.01 to 32.03 nm, with an average diameter of 16.92 nm.

ACKNOWLEDGEMENT

The researcher would like to thank all members of the Research Center for Laser and Advanced Nanotechnology (RC-LAN) for the assistance and support provided.

REFERENCES

- [1] Manukumar, N., Kishore, B., Manjunath, K., Nagaraju, G. 2018. Mesoporous Ta₂O₅ Nanoparticles as an Anode Material for Lithium Ion battery and an Efficient Photocatalyst for Hydrogen Evolution. *International Journal of Hydrogen Energy*. 43, 39, 18125-18135. <https://doi.org/10.1016/j.ijhydene.2018.08.075>.
- [2] Bae, Y.W., Lee, W.Y., Stinton, D.P. 1995. Effects of Temperature and Reagent Concentration on the Morphology of Chemically Vapor Deposited B-Ta₂O₅. *J. Am. Ceram. Soc.* 78, 1297–1300. <https://doi.org/10.1111/j.1151-2916.1995.tb08485.x>.
- [3] Choi, G.M., Tuller, H.L., Haggerty, J.S. 1989. Alpha - Ta₂O₅: An Intrinsic Fast Oxygen Ion Conductor. *J. Electrochem. Soc.* 136, 835–838. <https://doi.org/10.1149/1.2096752>.
- [4] Audier, M., Chenevier, B., Roussel, H., Vincent, L., Salaun, A.L. 2011. A Very Promising Piezoelectric Property of Ta₂O₅ thin Film: Monoclinic-trigonal Phase Transition, *journal of Solid state Chemistry*, 184 (8), 2023-2032. <https://doi.org/10.1016/j.jssc.2011.06.001>.
- [5] Sreethawong, T., Ngamsinlapasathian, S., Suzuki, Y., Yoshikawa, S. 2005. Nanocrystalline Mesoporous Ta₂O₅-Based Photocatalysts Prepared by Surfactant-Assisted Templating Sol–Gel Process for Photocatalytic H₂ Evolution. *J. Mol. Catal. A-Chem.* 235, 1–11. <https://doi.org/10.1016/j.molcata.2005.03.021>.
- [6] Ullah, R., Sun, H.Q., Ang, H.M., Tadé, M.O., Wang, S.B. 2013. Comparative Investigation of Photocatalytic Degradation of Toluene on Nitrogen Doped Ta₂O₅ and Nb₂O₅ Nanoparticles. *Industrial & Engineering Chemistry Research*. 52, 9, 3320–3328. <https://doi.org/10.1021/ie302326h>.
- [7] Wang, Y., Cui, Z., Zhang, Z. 2004. Synthesis and Phase Structure of Tantalum Nanoparticles. *Materials Letters*. 58, 3017 – 3020. <https://doi.org/10.1016/j.matlet.2004.05.031>.

- [8] Sreethawong, T., Ngamsinlapasathian, S., Yoshikawa, S. 2013. Facile Surfactant-Aided Sol–Gel Synthesis of Mesoporous-Assembled Ta₂O₅ Nanoparticles with Enhanced Photocatalytic H₂ Production. *Journal of Molecular Catalysis A: Chemical*. 374–375, 94–101. <http://dx.doi.org/10.1016/j.molcata.2013.04.003>.
- [9] Nagaraju, G., Karthik, K., Shashank, M. 2019. Ultrasound-Assisted Ta₂O₅ Nanoparticles and Their Photocatalytic and Biological Applications. *Journal Microchemical*. 147, 749-754. <https://doi.org/10.1016/j.microc.2019.03.094>.
- [10] Gomes, L.E., Silva, M.F., Goncalves, R/V., Machado, G., Alcantara, G.B., Caires, A., Wender, H. 2018. Synthesis and Visible-Light-Driven Photocatalytic Activity of Ta⁴⁺ Self-Doped Gray Ta₂O₅ Nanoparticles. *The Journal of Physical Chemistry C*. 122 (11), 6014-6025. <https://doi.org/10.1021/acs.jpcc.7b11822>.
- [11] Kim, M., Osone, S., Kim, T., Higashi, H., Seto, T. 2017. Synthesis of Nanoparticles by Laser Ablation: A Review. *KONA powder and particle journal*. 34, 80-90. <https://doi.org/10.14356/kona.2017009>.
- [12] Lu, QF, Li, JL, Yu, J., Cui, LJ, Wang, B., Ma, XJ, Feng, Y. 2021. Preparation of Ta₂O₅ Nanoparticles by Using Cathode Glow Discharge Electrolysis. *Mater. Res. Express* 8 125011. <https://doi.org/10.1088/2053-1591/ac3e94>.